

Hexaamminecobalt(III) hexacyanido-manganate(III)

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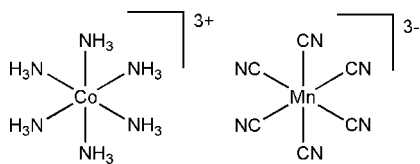
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 Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{N}-\text{C}) = 0.003$ Å; R factor = 0.025; wR factor = 0.071; data-to-parameter ratio = 19.6.

The asymmetric unit of the title compound, $[\text{Co}(\text{NH}_3)_6][\text{Mn}(\text{CN})_6]$, contains one Co and one Mn atom, both lying on threefold inversion axes, and one NH_3 and one CN group. The octahedral environments around Co^{III} and Mn^{II} are generated by symmetry and show very slight deviations from ideal geometry. A three-dimensional network is created by $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds.

Related literature

For related structures, see: Buschmann *et al.* (1999). For the construction of clusters and networks with adjustable magnetic properties, see: Przychodzen *et al.* (2006); Withers *et al.* (2005).



Experimental

Crystal data

 $[\text{Co}(\text{NH}_3)_6][\text{Mn}(\text{CN})_6]$
 $M_r = 372.19$

 Trigonal, $R\bar{3}$
 $a = 10.963$ (5) Å

 $c = 10.779$ (5) Å

 $V = 1121.9$ (9) Å³
 $Z = 3$

 Mo $K\alpha$ radiation

 $\mu = 1.96$ mm⁻¹
 $T = 100$ (2) K

 $0.35 \times 0.26 \times 0.25$ mm

Data collection

 Bruker SMART APEXII CCD
 area-detector diffractometer
 Absorption correction: multi-scan
 (SADABS; Bruker, 2004)
 $T_{\text{min}} = 0.546$, $T_{\text{max}} = 0.614$

 3602 measured reflections
 628 independent reflections
 497 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.038$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.025$
 $wR(F^2) = 0.071$
 $S = 0.93$
 628 reflections

 32 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.29$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.43$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1A}\cdots\text{N2}^{\text{i}}$	0.89	2.09	2.979 (2)	173

 Symmetry code: (i) $-y + 1, x - y + 1, z$.

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT-Plus (Bruker, 2004); data reduction: SAINT-Plus and XPREP (Bruker, 2004); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: DIAMOND (Brandenburg & Putz, 2005); software used to prepare material for publication: SHELXL97 and WinGX (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: PK2122).

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supplementary materials

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Hexaamminecobalt(III) hexacyanidomanganate(III)

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Comment

Our interest is in the use of cyanometalates as molecular building blocks for potentially constructing clusters and networks with adjustable magnetic properties [Przychodzen *et al.*, (2006), Withers *et al.*, (2005)].

The title compound crystallizes in the trigonal $R\bar{3}$ space group with $Z = 3$. The main part of the asymmetric unit contains one Co and one Mn atom, both lying on threefold rotational axes.

The octahedral environments around Co^{II} and Mn^{II} are generated by symmetry and shows very slight deviation from ideal geometry as illustrated by the C—Mn—C angles of 180.00 (7), 89.80 (7) and 90.20 (7) and N—Co—N angles of 180.00 (7), 89.47 (6) and 90.53 (6) ° respectively.

A three dimensional network is created by hydrogen bonds of the type N—H—N.

Experimental

Equimolar amounts of $\text{K}_3[\text{Mn}(\text{CN})_6]$ and $[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ were dissolved in water, added together and allowed to stand. Orange crystals separated out after a few days.

Figures



Fig. 1. An ellipsoid plot of the title compound (50% probability displacement ellipsoids).

Hexaamminecobalt(III) hexacyanidomanganate(III)

Crystal data

$[\text{Co}(\text{NH}_3)_6][\text{Mn}(\text{CN})_6]$

$M_r = 372.19$

Trigonal, $R\bar{3}$

Hall symbol: $-R\ 3$

$a = 10.963$ (5) Å

$b = 10.963$ (5) Å

$c = 10.779$ (5) Å

$\alpha = 90^\circ$

$Z = 3$

$F_{000} = 570$

$D_x = 1.653$ Mg m $^{-3}$

Mo $K\alpha$ radiation

$\lambda = 0.71069$ Å

Cell parameters from 1532 reflections

$\theta = 2.9\text{--}28.3^\circ$

$\mu = 1.96$ mm $^{-1}$

$T = 100$ (2) K

supplementary materials

$\beta = 90^\circ$
 $\gamma = 120^\circ$
 $V = 1121.9 (9) \text{ \AA}^3$

Cuboid, orange
 $0.35 \times 0.26 \times 0.25 \text{ mm}$

Data collection

Bruker SMART APEXII CCD area-detector
diffractometer
 $R_{\text{int}} = 0.038$

φ and ω scans
 $\theta_{\text{max}} = 28.3^\circ$

Absorption correction: multi-scan
(SADABS; Bruker, 2004)
 $\theta_{\text{min}} = 2.9^\circ$

$T_{\text{min}} = 0.546$, $T_{\text{max}} = 0.614$
 $h = -12 \rightarrow 14$

3602 measured reflections
 $k = -14 \rightarrow 9$

628 independent reflections
 $l = -10 \rightarrow 14$

497 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2
Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.025$
 $wR(F^2) = 0.071$
 $S = 0.93$
628 reflections
32 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0379P)^2 + 3.8029P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.29 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.42 \text{ e \AA}^{-3}$
Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
Co1	0	0	0	0.00732 (17)
Mn1	0.3333	0.6667	0.1667	0.00607 (17)
N1	-0.02748 (15)	0.13048 (15)	-0.10659 (13)	0.0107 (3)
H1A	0.0238	0.2179	-0.0777	0.016*
H1B	-0.1181	0.1061	-0.1065	0.016*
H1C	-0.0006	0.1264	-0.1837	0.016*
C2	0.31385 (18)	0.80238 (18)	0.06080 (16)	0.0119 (3)
N2	0.30018 (17)	0.87891 (17)	-0.00283 (15)	0.0184 (4)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Co1	0.0076 (2)	0.0076 (2)	0.0067 (3)	0.00381 (10)	0	0
Mn1	0.0064 (2)	0.0064 (2)	0.0055 (3)	0.00318 (10)	0	0
N1	0.0117 (7)	0.0099 (7)	0.0111 (7)	0.0058 (6)	0.0007 (5)	0.0010 (5)
C2	0.0108 (8)	0.0144 (8)	0.0115 (8)	0.0070 (7)	0.0001 (6)	-0.0026 (6)
N2	0.0200 (8)	0.0248 (9)	0.0159 (7)	0.0152 (7)	0.0014 (6)	0.0013 (6)

Geometric parameters (\AA , $^\circ$)

Co1—N1 ⁱ	1.9718 (16)	Mn1—C2 ^{vii}	1.9696 (19)
Co1—N1 ⁱⁱ	1.9718 (16)	Mn1—C2 ^{viii}	1.9696 (19)
Co1—N1 ⁱⁱⁱ	1.9718 (15)	Mn1—C2 ^{ix}	1.9696 (19)
Co1—N1 ^{iv}	1.9718 (15)	Mn1—C2 ^x	1.9696 (19)
Co1—N1	1.9718 (15)	N1—H1A	0.89
Co1—N1 ^v	1.9718 (15)	N1—H1B	0.89
Mn1—C2 ^{vi}	1.9696 (19)	N1—H1C	0.89
Mn1—C2	1.9696 (19)	C2—N2	1.150 (2)
N1 ⁱ —Co1—N1 ⁱⁱ	180.00 (7)	C2—Mn1—C2 ^{viii}	89.80 (7)
N1 ⁱ —Co1—N1 ⁱⁱⁱ	89.47 (6)	C2 ^{vii} —Mn1—C2 ^{viii}	89.80 (7)
N1 ⁱⁱ —Co1—N1 ⁱⁱⁱ	90.53 (6)	C2 ^{vi} —Mn1—C2 ^{ix}	89.80 (7)
N1 ⁱ —Co1—N1 ^{iv}	89.47 (6)	C2—Mn1—C2 ^{ix}	90.20 (7)
N1 ⁱⁱ —Co1—N1 ^{iv}	90.53 (6)	C2 ^{vii} —Mn1—C2 ^{ix}	180
N1 ⁱⁱⁱ —Co1—N1 ^{iv}	89.47 (6)	C2 ^{viii} —Mn1—C2 ^{ix}	90.20 (7)
N1 ⁱ —Co1—N1	90.53 (6)	C2 ^{vi} —Mn1—C2 ^x	89.80 (7)
N1 ⁱⁱ —Co1—N1	89.47 (6)	C2—Mn1—C2 ^x	90.20 (7)
N1 ⁱⁱⁱ —Co1—N1	180.00 (10)	C2 ^{vii} —Mn1—C2 ^x	90.20 (7)
N1 ^{iv} —Co1—N1	90.53 (6)	C2 ^{viii} —Mn1—C2 ^x	180.00 (7)
N1 ⁱ —Co1—N1 ^v	90.53 (6)	C2 ^{ix} —Mn1—C2 ^x	89.80 (7)
N1 ⁱⁱ —Co1—N1 ^v	89.47 (6)	Co1—N1—H1A	109.5
N1 ⁱⁱⁱ —Co1—N1 ^v	90.53 (6)	Co1—N1—H1B	109.5
N1 ^{iv} —Co1—N1 ^v	180.00 (10)	H1A—N1—H1B	109.5
N1—Co1—N1 ^v	89.47 (6)	Co1—N1—H1C	109.5
C2 ^{vi} —Mn1—C2	180	H1A—N1—H1C	109.5
C2 ^{vi} —Mn1—C2 ^{vii}	90.20 (7)	H1B—N1—H1C	109.5
C2—Mn1—C2 ^{vii}	89.80 (7)	N2—C2—Mn1	178.31 (17)
C2 ^{vi} —Mn1—C2 ^{viii}	90.20 (7)		

Symmetry codes: (i) $y, -x+y, -z$; (ii) $-y, x-y, z$; (iii) $-x, -y, -z$; (iv) $x-y, x, -z$; (v) $-x+y, -x, z$; (vi) $-x+2/3, -y+4/3, -z+1/3$; (vii) $-y+1, x-y+1, z$; (viii) $-x+y, -x+1, z$; (ix) $y-1/3, -x+y+1/3, -z+1/3$; (x) $x-y+2/3, x+1/3, -z+1/3$.

supplementary materials

Hydrogen-bond geometry (Å, °)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N1-H1A\cdots N2^{vii}$	0.89	2.09	2.979 (2)	173

Symmetry codes: (vii) $-y+1, x-y+1, z$.

Fig. 1

